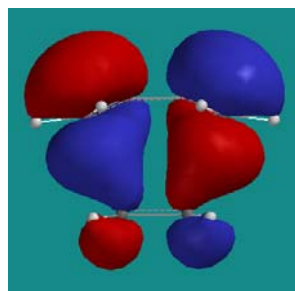


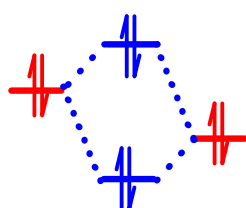
MO Compost

- Functional group \rightarrow **atomic orbitals**
- Molecule w/ extended π system \rightarrow **functional group orbitals**
- Bimolecular \ddagger \rightarrow **molecular orbitals**



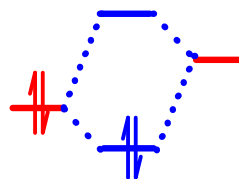
butadiene + ethylene HO-1

Key MO Combinations in Transition State



2 filled **Rct** MOs mix
make 2 filled **TS** MOs

net destabilization



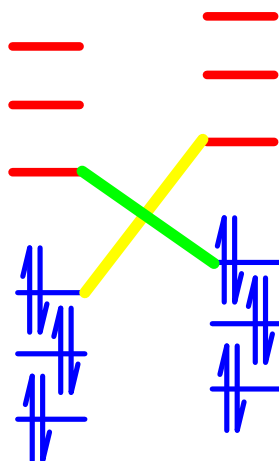
filled + empty **Rct** MOs mix
make filled + empty **TS** MOs

net stabilization

Frontier MO

- Lower $E(\ddagger)$ due to stabilizing MO interactions
- Stabilization $E \approx \mu = (H_{AB})^2 / (E_B - E_A)$
 - Many possible pairings & stabilizations
- Which do we consider?
 - Largest stabilization from HOMO-LUMO
 - minimizes $E_B - E_A$
 - Evaluate 2 HOMO-LUMO
 - “dominant” interaction minimizes $E_B - E_A$

Gaps & Rxn Polarity



- e-rich molecule
 - high HOMO / LUMO
- e-poor molecule
 - low HOMO / LUMO
- dominant FMO =
 - $\text{HOMO}_{\text{rich}}$ & $\text{LUMO}_{\text{poor}}$
- polar rxns fast
 - small gap
 - delocalize electrons
 - $\text{HO}_{\ddagger} \approx \text{HO}_{\text{rich}} + \lambda \text{LU}_{\text{poor}}$

Gaps & Rates

- Normal Diels-Alder
 - e-rich diene (high HOMO)
 - e-poor dienophile (low LUMO)
- Dienophile LUMO
 - bonding interaction w/ π system of X stabilizes LUMO

| CH ₂ =CHX | LUMO (eV) |
|---------------------------------|-----------|
| OCH ₃ | 5.7 |
| CH ₂ CH ₃ | 5.2 |
| H | 5.1 |
| SCH ₃ | 4.3 |
| CF ₃ | 3.6 |
| CH=CH ₂ | 3.6 |
| SO ₂ CH ₃ | 3.1 |
| C≡N | 2.8 |
| CH=O | 2.6 |
| NO ₂ | 1.3 |

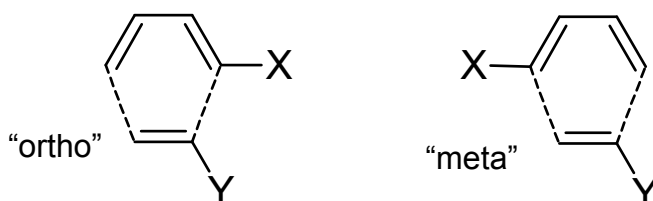
Gaps & Rates

- Inverse Electron Demand Diels-Alder
 - e-poor diene (low LUMO)
 - e-rich dienophile (high HOMO)
- Dienophile HOMO
 - antibonding interaction w/ π system of X destabilizes HOMO

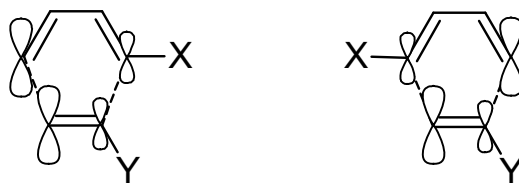
| CH ₂ =CHX | HOMO (eV) |
|---------------------------------|--------------|
| SCH ₃ | -8.8 |
| CH=CH ₂ | -8.8 |
| OCH ₃ | -9.3 |
| CH ₂ CH ₃ | -9.8 |
| H | -10.3 |
| CH=O | -10.7 |
| C≡N | -10.8 |
| SO ₂ CH ₃ | -11.5 |
| CF ₃ | -11.6 |
| NO ₂ | -12.0 (HO-1) |

Overlap & Regioselectivity

- Identify dominant FMO interaction
 - Same $E_B - E_A$ for both transition states
- Estimate **overlap** for 2 transition states
 - Largest stabilization from best FMO overlap



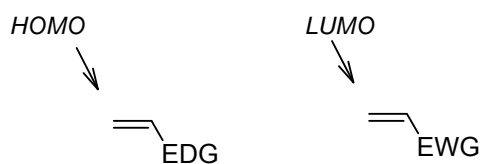
Overlap Simplified



- FMO sweep over several atoms
 - look at overlap only where reactants approach
- Total overlap \approx sum of "pair overlaps"
 - best total overlap matches "largest" part of HOMO_B with "largest" part of LUMO_A
 - example in drawing, ortho overlap > meta overlap

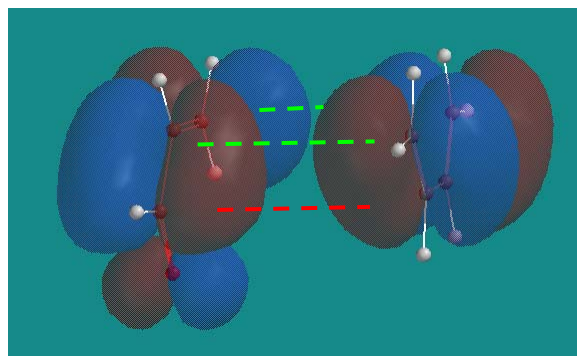
FMO Shapes

- Pencil & Paper
 - Do “mini” MO calculation (QM required)



- Computer
 - MO surface ☹ vs. MO coefficients ☺

Endo Rule – Secondary MO Interactions?



additional “pair overlap” enhances H_{AB}
and stabilizes \ddagger